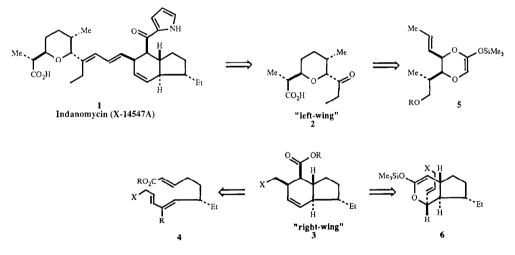
NOT THE ENOLATE CLAISEN REARRANGEMENT. A SURPRISING ROUTE TO THE "RIGHT-WING" OF INDANOMYCIN (X-14547A)

Steven D. Burke,*¹ David M. Armistead, and K. Shankaran Department of Chemistry, University of South Carolina Columbia, South Carolina 29208

Abstract: A synthesis of the "right-wing" equivalent 22 of the ionophore antibiotic indanomycin is described, wherein an unexpected retro hetero Diels-Alder/intramolecular Diels-Alder pathway gives the desired product of the planned Claisen [3,3]-shift.

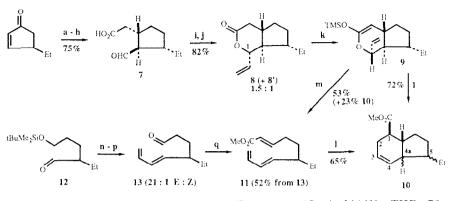
The ionophore antibiotic indanomycin (X-14547A, 1)² can be divided into synthetic subgoals 2 and 3, respectively described as the "left-wing" and "right-wing". An intramolecular Diels-Alder approach to the hydrindene subunit 3 from a suitable trienoic ester such as 4 has been amply demonstrated.^{2m-p} The recognition of 3 as a γ , δ -unsaturated ester suggested that a Claisen rearrangement³ strategy might provide an alternative route to this "right-wing". Since we had previously developed⁴ an efficient synthesis of the "left-wing" hydropyran 2 via a Claisen rearrangement of the silyl ketene acetal 5, the possibility of assembling each structurally dissimilar subunit 2 and 3 via a common method was attractive.



In accord with available precedent⁵ for the lactone enolate Claisen rearrangement in the synthesis of cyclohexene carboxylic acids, the construction of 3 would require that the silyl ketene acetal 6 undergo [3,3] sigmatropic reorganization via a boat-like transition state, as shown. However, the precedent did not extend to bicyclic systems with a *trans*-fused cyclopentane ring, and examination of molecular models of 6 revealed that even modest pericyclic overlap would entail substantial ring strain. Thus, a somewhat simpler system than 6 was initially chosen to test the feasibility of this questionable step.

In a pedestrian but high-yielding sequence of reactions, 4-ethylcyclopentenone⁶ was converted to the aldehyde 7 (Scheme I). Addition of this aldehyde to excess vinylmagnesium bromide in tetrahydrofuran (THF) at -78°C gave in 85% yield a mixture of allylic alcohols. Lactonization⁷ with dicyclohexylcarbodiimide (DCC) and 4-(N,N-dimethylamino)pyridine (DMAP) provided in 96% yield the C(1)-epimeric δ -lactones 8 (mp 36-38°C) and 8', separable by flash chromatography,⁸ in a ratio of 1.5:1.

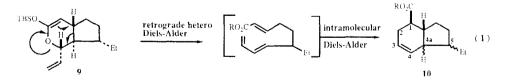




(a) $(H_2C = CH)_2CuLi$, THF, -78°C; BrCH₂CO₂t-Bu, HMPA, -78 \rightarrow 0°C. (b) LiAlH₄, THF, -78 \rightarrow 25°C. (c) *t*-BuPh₂SiCl, DMF, imidazole, 25°C. (d) MsCl, DMAP, Et₃N, CH₂Cl₂, 0 \rightarrow 25°C. (e) LiEt₃BH, THF, reflux. (f) *n*-Bu₄NF, THF, 25°C. (g) H₂Cr₂O₇, aq. acetone, 25°C. (h) O₃, 1:1 CH₂Cl₂-MeOH, -78°C. (i) H₂C = CHMgBr, THF, -78°C. (j) DCC, DMAP, CH₂Cl₂, 0 \rightarrow 25°C.⁷ (k) LiN(SiMe₃)₂, Me₃SiCl, Et₃N, THF, -100 \rightarrow 25°C; remove THF *in vacuo*, add PhCH₃. (l) 135°C, 20 h; aq. HCl; CH₂N₂, Et₂O, 0°C. (m) 95-100°C, 5 h; aq. HCl; CH₂N₂, Et₂O, 0°C. (n) Ph₂P(O)CHLiCH = CH₂, THF, -78 \rightarrow 25°C.²⁰ (o) *n*-Bu₄NF, THF, 25°C. (p) CrO₃ • 2 pyr, CH₂Cl₂, 0 \rightarrow 25°C. (q) Ph₃PCHCO₂Me, CH₂Cl₂, 25°C.

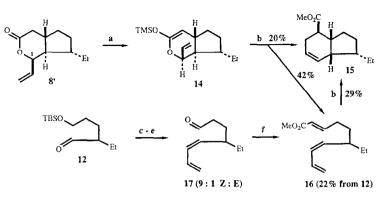
Conversion of lactone 8 to the trimethylsilyl ketene acetal 99 and thermolysis thereof in toluene at 135°C (bath temperature) gave in 72% yield the hydrindene nucleus 10 as a mixture of four diastereomers (31:9:5:1).¹⁰ The major product [a-H at C(4a), a-Et at C(5)] was that expected from a Claisen [3,3]-sigmatropic reorganization of 9. However, the formation of additional diastereomers was indicative of another pathway, revealed by conducting the thermolysis under milder conditions (95-100°C, 5 h).¹¹ In addition to the hydrindenes 10 (23%), there was isolated the *E.E.*-triene 11 in 53% yield. Resubjection of 11 to thermolysis at 135°C for 20 h gave in 65% yield the hydrindene mixture 10. Confirmation of the structure of triene 11 was accomplished by independent synthesis from aldehyde 12^{2m} ,0 via dienal 13 as shown.

It was thus apparent that a retrograde hetero Diels-Alder reaction was preempting the desired [3,3]shift, but that a subsequent intramolecular Diels-Alder¹² process of higher activation energy led to the expected product [C(4a) α -H, C(5) α -Et] as the major diastereomer (eq. 1). The presence of the three minor cycloadducts was a consequence of the imperfect stereoselectivity of the Diels-Alder conversion to 10.13



As shown in Scheme II, similar results were obtained upon thermolysis of the silvl ketene acetal 14 derived from 8'. In this case a single hydrindene, to which we assign structure $15,^{14}$ was produced (20%) in addition to the *E*,*Z*-triene 16 (42%). The high stereoselectivity in the Diels-Alder stage of this cycloreversion/cycloaddition is attributable to the relative unreactivity of the triene 16,¹⁵ The structure of 16 was confirmed by independent synthesis from 12 as shown, and thermolysis of this material gave the same hydrindene as had 14 under like conditions.

SCHEME II¹⁹

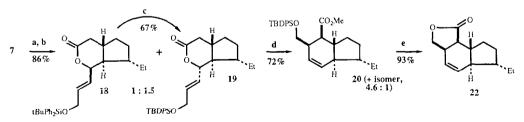


(a) LiN(SiMe₃)₂, Me₃SiCl, Et₃N, THF, -100 \rightarrow 25°C; remove THF *in vacuo*, add PhCH₃. (b) 135°C, 20 h; aq. HCl; CH₂N₂, Et₂O, 0°C. (c) Ph₂PCHLiCH = CH₂, Ti(Oi-Pr)₄, THF, -78°C; MeI, -78 \rightarrow 25°C.²⁰ (d) *n*-Bu₄NF, THF, 25°C. (e) CrO₃ • 2 pyr, CH₂Cl₂, 0 \rightarrow 25°C. (f) Ph₃PCHCO₂Me, CH₂Cl₂, 25°C.

Our doubts about the viability of the Claisen rearrangement route to the "right-wing" of indanomycin had thus been confirmed. However, the intervening tandem sequence of retro hetero Diels-Alder/intramolecular Diels-Alder reactions gave the desired product in the model conversion $(8 \rightarrow 10)$. Therefore, the intended application was pursued.

Addition of (*E*)-3-bromomagnesio-1-*t*-butyldiphenylsiloxy-2-propene¹⁶ to the aldehyde **7** gave a mixture of hydroxy acids (94%) which, upon lactonization with DCC/DMAP,⁷ gave a 91% yield of the separable lactones **18** and **19** (mp 79-80°C) in a ratio of 1:1.5 (Scheme III).¹⁷ The minor, undesired isomer **18** was converted to **19** by hydrolysis and relactonization by the Mitsunobu procedure¹⁸ in 67% overall yield. Conversion of **19** to the trimethylsilyl ketene acetal⁹ and thermolysis in toluene at 135°C (bath temperature) for 24 h gave, following hydrolysis and esterification, a 72% yield of the known.²⁰ *trans*-fused hydrindene **20** and a separable diastereomer in a ratio of 4.6:1. Removal of the silyl protecting group with *n*-Bu₄NF as previously described²⁰ proceeded with lactonization to give in 93% yield the known "right-wing" synthon **22** (mp 68-69°C; lit. 68-68.5°C,²⁰ 67.5-68.5²m), identical with that described by Nicolaou²⁰ and by Ley.^{2m} The overall yield for the production of **22** from 4-ethylcyclopentenone exceeds 20%. This work, when combined with previous efforts from our labs⁴ and others,^{2m,0} constitutes a formal total synthesis of indanomycin (X-14547A, 1).

SCHEME III¹⁹



(a) (E)-t-BuPh₂SiOCH₂CH = CHMgBr,¹⁶ THF, -78°C. (b) DCC, DMAP, CH₂Cl₂, $0 \rightarrow 25^{\circ}$ C.7 (c) LiOH, aq. THF, 25°C; 5% aq. HCl; DEAD, Ph₃P, PhCH₃, -20°C.¹⁸ (d) LDA, Me₃SiCl, Et₃N, THF, -100 $\rightarrow 25^{\circ}$ C; remove THF *in vacuo*, add PhCH₃; 135°C, 24 h; aq. HCl; CH₂N₂, Et₂O, 0°C. (e) *n*-Bu₄NF, THF, 0°C, 2.5 h.²⁰

Acknowledgement. We gratefully acknowledge the National Institutes of Health, the Alfred P. Sloan Foundation, and the National Science Foundation for generous financial support. Industrial matching

6298

funds for the NSF Presidential Young Investigator Award from Stuart Pharmaceuticals, Rohm and Haas Co., DuPont, Union Camp, SOHIO, and Hardwicke Chemicals are greatly appreciated. Support of high-field NMR spectrometer purchases at the University of South Carolina by the NSF (CHE 82-07445. CHE 84-11172) and the NIH (1S10 RRO2425) is acknowledged. We are grateful to Professor Steven Ley (Imperial College) for providing spectra of 22 for comparison.

References and Notes

- Recipient of a National Science Foundation Presidential Young Investigator Award: Research 1. Fellow of the Alfred P. Sloan Foundation.
- 2. Isolation, structural and biological characterization: (a) Westley, J. W.; Evans, R. H., Jr.; Liu, C.-M.; Hermann, T.; Blount, J. F. J. Am. Chem. Soc. 1978, 100, 6786. (b) Liu, C.-M.; Hermann, T. E.; Liu, M.; Bull, D. N.; Palleroni, N. J.; Prosser, B. L. T.; Westley, J. W.; Miller, P. A. J. Antibiot. 1979, 32, 95. (c) Westley, J. W.; Evans, R. H., Jr.; Sello, L. H.; Troupe, N.; Liu, C.-M.; Blount, J. F. Ibid. 1979, 32, 100. (d) Westley, J. W.; Liu, C.-M. U.S. Patent 4100171, 1978. For other synthetic efforts directed at X-14547A, see: (e) Nicolaou, K. C.; Magolda, R. L. J. Org. Chem. 1981, 46, 1506. (f) Roush, W. R.; Myers, A. G. J. Org. Chem. 1981, 46, 1509. (g) Edwards, M. P.; Ley, S. V.; Lister, S. G. Tetrahedron Lett. 1981, 22, 361. (h) Nicolaou, K. C.; Papahatjis, D. P.; Claremon, D. A.; Papahatiis, D. P.; Magolda, R. L. Ibid. 1981, 103, 6969. (i) Ho P. Can. J. Chem. Isolation, structural and biological characterization: (a) Westley, J. W.; Evans, R. H., Jr.; Liu, C.-Claremon, D. A.; Papahatjis, D. P.; Magolda, R. L. Ibid. 1981, 103, 6969. (j) Ho. P. Can. J. Chem. 1982, 60, 90. (k) Roush, W. R.; Peseckis, S. M. Tetrahedron Lett. 1982, 23, 4879. (l) Edwards, M. P.; Ley, S. V.; Lister, S. G.; Palmer, B. D. J. Chem. Soc., Chem. Commun. 1983, 630. (m) Edwards, M. P.; Ley, S. V.; Lister, S. G.; Palmer, B. D. J. Chem. Sol., Chem. Commun. 1363, 6569, 4449 S503. (n) Roush, W. R.; Peseckis, S. M.; Walts, A. E. Ibid. 1984, 49, 3429. (o) Nicolaou, K. C.; Papahatjis, D. P., Claremon, D. A.; Magolda, R. C.; Dolle, R. E. Ibid. 1985, 50, 1440. (p) Boeckman, R. K., Jr.; Barta, T. E. J. Org. Chem. 1985, 50, 3423. (q) Whitney, R. A. Can. J. Chem. 1986, 64, 803.
- 3. (a) Ireland, R. E.; Daub, J. P. J. Org. Chem. 1981, 46, 479. (b) Ireland, R. E.; Vevert, J.-P. Ibid. 1980, 45, 4259. (c) Ireland, R. E.; Thaisrivongs, S.; Vanier, N.; Wilcox, C. S. Ibid. 1980, 45, 48. (d) Ireland, R. E.; Thaisrivongs, S.; Wilcox, C. E. J. Am. Chem. Soc. 1980, 102, 1155. (e) Ireland, R. E.; Mueller, R. H.; Willard, A. K. Ibid. 1976, 98, 2868. (f) Ireland, R. E.; Mueller, R. H. Ibid. 1972, 94, 5897.
- 4. (a) Burke, S. D.; Armistead, D. M.; Schoenen, F. J.; Fevig, J. M. Tetrahedron 1986, 42, 2787. (b) Burke, S. D.; Armistead, D. M.; Fevig, J. M. Tetrahedron Lett. 1985, 26, 1163.
- 5. Danishefsky, S.; Funk, R. L.; Kerwin, J. F., Jr. J. Am. Chem. Soc. 1980, 102, 6889.
- 6. Prepared in 78% yield from trans-2-pentenoyl chloride and vinyltrimethylsilane, according to;
- Kipeldsen, G.; Knudsen, J. S.; Ravin-Peterson, L. S.; Torssell, K. B. G. Tetrahedron 1983, 39, 2237.
 (a) Hassner, A.; Alexanian, V. Tetrahedron Lett. 1978, 4475.
 (b) Neises, B.; Steglich, W. Angew. Chem., Int. Ed. Engl. 1978, 17, 522.
 (c) Ziegler, F. E.; Berger, G. D. Synth. Commun. 1979, 9, 539.
 Still, W. C.; Kahn, M.; Mitra, A. J. Org. Chem. 1978, 43, 2923. 7.
- 8.
- 9. In order to avoid enolate fragmentation, the lactone was deprotonated with LDA or LHMDS, as indicated, in the presence of Me₃SiCl at -100°C. See: Corey, E. J.; Gross, A. W. Tetrahedron Lett. 1984, 25, 495.
- 10. The ratio was determined by capillary GC at 130°C on a 25 m column coated with SUPEROX-4 (Alltech Associates, Deerfield, IL).
- 11. The multiple diastereomeric products clearly implicated a temporary loss of sp3-carbon stereo-
- genicity in the course of the conversion of 8 to 10. For recent reviews of the intramolecular Diels-Alder reaction, see: (a) Ciganek, E.; Org. Reac-12. tions 1985, 32, 1. (b) Taber, D. F. in *Reactivity and Structure Concepts in Organic Chemistry*, Springer Verlag, Berlin and New York, 1984. (c) Fallis, A. G. *Can. J. Chem.* 1984, 62, 183. For related observations in a systematic study, see: Roush, W. R.; Gillis, H. R.; Ko, A. I. *J. Am.*
- 13. Chem. Soc. 1982, 104, 2269.
- 14. The product 15 was not unambiguously characterized; the stereochemical assignment is based upon analogy with the observations of Roush¹³ and House.¹⁵
- 15. House, H. O.; Cronin, T. H. J. Org. Chem. 1965, 30, 1061.
- 16. Prepared by transmetalation [*n*-BuLi, THF, $-78 \rightarrow -20^{\circ}$ C; MgBr₂ (1 M in 3:1 Et₂O-PhH), -78° C] of the t-butyldiphenylsilyl derivative of the known E-3-(tri-n-butyl)stannyl-2-propen-1-ol. See: Jung, M. E.; Light, L. A. Tetrahedron Lett. 1982, 23, 3851.
- These isomers are readily distinguished by their vicinal couplings between the angular and carbinyl protons in the ¹H NMR spectra at 400 MHz. 17.
- 18. Mitsunobu, O. Synthesis 1981, 1.
- 19. Yield cited in the Schemes are for chromatographically and spectroscopically pure substances, except where mixtures are described. All structural assignments are supported by IR, high-field ¹H NMR, ¹³C NMR, and mass spectrometric and elemental analyses. All chiral substances were produced as racemates; a single enantiomer is shown for simplicity.
- 20. Ukai, J.; Ikeda, Y.; Ikeda, N.; Yamamoto, H. Tetrahedron Lett. 1983, 24, 4029.